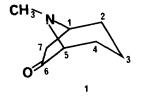
## 6-β-ARYLTROPAN-6-OLS - A NEW SERIES OF POTENTIAL ANALGESICS

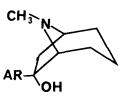
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The chemistry and pharmacological properties of naturally occurring tropanes such as hyoscine, hyoscyamine and cocaine have undergone extensive investigation and as a consequence many synthetic antispasmodics and local anaesthetics have been developed. Analgesics related to pethidine, which owes its origins to atropine studies and based upon 3-substituted tropanes have also been synthesised and activities greater than pethidine have been reported (Casy 1978).

In contrast to 3-substituted tropanes, compounds with substituents in the 6-position are sterically constrained and there is a fixed spatial relationship between the tropane nitrogen and  $\alpha-$  or  $\beta-$  6 substituents. This offers a potential tool for the study of pharmacophore-receptor interactions in novel neurotransmitter mimics and antagonists. 6-Aryltropans exhibit structural similarities to 3-arylpiperidines (Iorio and Casy 1978) and 3-arylpyrrolidines (Bowman et al 1973), members of which groups show analgesic agonist and antagonist activities over a wide dose range.

Neglect of 6-substituted tropanes is a result of synthetic difficulties particularly on scale-up. The route to the key intermediate tropan-6-one(1) follows a classical Robinson reaction where hydroxysuccinaldehyde, prepared in situ is condensed with methylamine and acetonedicarboxylic acid in citrate buffer at pH5, to give  $6\beta$ -hydroxytropan-3-one. The 3-carbonyl group is removed by Wolff-Kishner reduction and chromic acid oxidation affords tropan-6-one.





Reaction of aryl Grignard reagents with tropan-6-one affords 6-aryltropan-6-ols(2) in good yield and a predominant 6 $\beta$ -aryl isomer. The orientation of the aryl ring may be predicted as occurring  $\beta$ - by a consideration of the Grignard/tropan-6-one transition complex. Confirmation of this came from lOOMHz  $^1$ H NMR studies of both the tertiary base of 6-phenyltropan-6-ol and its methiodide. From the spectra it is clear that one quaternary +N.CH3 signal occurs at the uncharacteristic high field resonance of  $^5$ 2.67. The only reasonable explanation for this is that the equatorial +N.CH3 group experiences significant anisotropic shielding from the phenyl ring with a  $\beta$ -orientation.

 $\underline{N}\text{-}\text{Demethylation}$  of  $6\beta\text{-}\text{phenyltropan-}6\text{-}\text{ol}$  was affected by trichloroethylchloroformate and agonist and antagonist substituents inserted by standard techniques. Propionyloxy- and acetyloxy-esters bearing a structural resemblance to reversed esters of pethidine were prepared.

Tropane esters bearing nitrogen substituents characteristic of agonists were low potency morphine-like analgesics on the guinea-pig ileum; agents with antagonist substituents were inactive in tests of reversal of morphine response.

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Casy, A.F., Progress in Drug Research 22, 149-227 (1978)
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